
How To Make Dakins Solution

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UNDERSTANDING DAKIN'S SOLUTION AND ITS USES

Dakin's solution is a dilute antiseptic made from sodium hypochlorite, initially developed during World War I by surgeon Henry Dakin and chemist Alexis Carrel. It became a staple in wound care because it effectively kills bacteria, including MRSA and Pseudomonas, while being less damaging to healthy tissue than full-strength bleach. The solution is typically buffered with either baking soda (sodium bicarbonate) or boric acid to maintain a neutral to slightly alkaline pH, which helps reduce stinging when applied to open wounds. Doctors often recommend Dakin's solution for cleaning chronic wounds, pressure ulcers, diabetic foot ulcers, and post-surgical incisions that are infected or at high risk of infection. However, it is not intended for deep wounds, internal use, or for prolonged application on healthy skin. Understanding **how to make Dakin's solution** correctly is critical because an improper concentration can either be ineffective or cause tissue irritation.

The core active ingredient in Dakin's solution is sodium hypochlorite, which is the same chemical found in household bleach. However, household bleach contains extra additives (such as surfactants, thickeners, or fragrances) that can damage delicate wound tissue. Therefore, you must use plain, unscented, non-thickened, 5.25%

sodium hypochlorite bleach for this recipe. Additionally, the water used should be distilled or sterile, as tap water contains minerals and chlorine that can cause the solution to degrade too quickly or leave harmful residues. Many pharmacies and compounding centers sell pre-made Dakin's solution, but for those who need it in a pinch or want to control freshness, making it at home is straightforward.

ESSENTIAL INGREDIENTS AND EQUIPMENT

Before you begin, gather the following items to ensure your solution is both safe and effective. Using the wrong type of bleach or water can lead to poor results or even increase infection risk.

- **Household bleach (5.25% sodium hypochlorite)** – do not use splash-less, scented, concentrated (8.25% or higher), or “color-safe” bleach. Check the label for the percentage. If the percentage is different, you must adjust the dilution ratio.
- **Distilled or purified water** – available at most grocery stores. Do not use tap water unless you boil it first and let it cool.
- **Baking soda (sodium bicarbonate)** – pure baking soda without any added aluminum or chemicals. This buffers the solution to a pH of 8-9, making it gentler on wounds.
- **Clean glass container with a tight lid** – plastic can absorb

chemicals and is harder to sterilize. A dark amber or blue glass bottle is best because light degrades sodium hypochlorite.

- **Measuring spoons and a liquid measuring cup** – accuracy is important. Use separate utensils for the bleach and baking soda to avoid cross-contamination.
- **pH test strips (optional)** – to confirm the solution is between pH 7.5 and 9.0. If it is too acidic (below pH 7), it can burn the wound; too alkaline (above pH 10) reduces efficacy.
- **Labels and a marker** – always date your solution and note the strength.

STEP-BY-STEP GUIDE: HOW TO MAKE DAKIN'S SOLUTION

The two most common strengths are 0.5% (full strength) and 0.25% (half strength). Your doctor or wound care nurse will tell you which is appropriate for your situation. For routine wound cleaning, many experts recommend starting with half strength because it still kills most common pathogens but causes less stinging and less damage to new tissue growth. Below are precise measurements for a 1-liter batch. You can adjust the batch size proportionally.

Full Strength (0.5% Sodium Hypochlorite) – Makes 1 Liter

1. Measure 1 liter of distilled water and pour it into a clean glass

pitcher or bottle.

2. Add 1.5 teaspoons (about 7.5 grams) of baking soda. Stir or shake gently until fully dissolved. This step ensures the solution is buffered.
3. Measure 95 milliliters (about 6½ tablespoons or ⅓ cup plus 1 tablespoon) of 5.25% plain bleach. Slowly pour the bleach into the water-baking soda mixture.
4. Stir gently with a clean stainless steel or plastic utensil. Do not shake vigorously; you want to avoid creating bubbles that trap air and degrade the solution.
5. If you have pH test strips, dip one into the solution. The strip should change to a color indicating pH 8 to 9. If the pH is lower, add a pinch more baking soda; if higher, add a few drops of distilled water.
6. Pour the finished solution into a dark glass bottle, label it with the date and strength ("0.5% Dakin's"), and store it in a cool, dark place away from direct sunlight.

Half Strength (0.25% Sodium Hypochlorite) – Makes 1 Liter

1. Repeat steps 1 and 2 from the full-strength method: dissolve 1.5 teaspoons of baking soda in 1 liter of distilled water.
2. Instead of 95 mL of bleach, add only 47.5 mL (approximately 3.2 tablespoons) of 5.25% bleach.

3. Stir gently, test pH if desired, and store in a dark glass bottle labeled "0.25% Dakin's".

Important tip: If you are using a brand of bleach with a different percentage, recalculate the volume. For example, if your bleach is 6%, use slightly less than the amounts above to avoid making the solution too strong. The formula is: required bleach volume (mL) = (desired percentage × total volume) / (bleach percentage). Always mix in the order of water first, then baking soda, then bleach, because adding bleach to water is safer than the reverse.

IMPORTANT SAFETY PRECAUTIONS

While Dakin's solution is remarkably safe for topical wound care when prepared correctly, there are several critical points you must follow to avoid injury or infection.

- **Never use scented, thickened, or "splash-free" bleach.** These often contain surfactants, polymers, or fragrances that can severely irritate a wound or even

cause allergic reactions. Only plain 5.25% household bleach (unscented) is suitable.

- **Always use distilled or previously boiled water.** Tap water may contain chlorine, chloramine, fluoride, and microorganisms that can compromise the solution's stability and sterility. Tap water also contains minerals that can combine with bleach to produce unwanted compounds.
- **Never mix Dakin's solution with other chemicals.** Combining it with vinegar, hydrogen peroxide, or alcohol releases toxic chlorine gas. Even mixing it with different wound cleansers can cause dangerous reactions. Use Dakin's solution alone.
- **Do not use on deep wounds, open body cavities, or near eyes and mucous membranes.** Dakin's solution is for topical application to the skin and superficial wounds only. If the wound extends into muscle, bone, or a body cavity, consult a